

(12) INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

(19) World Intellectual Property  
Organization  
International Bureau



(43) International Publication Date  
4 March 2004 (04.03.2004)

PCT

(10) International Publication Number  
**WO 2004/017747 A1**

- (51) International Patent Classification<sup>7</sup>: A23G 9/02, 3/00, 9/04, 9/20
- (21) International Application Number: PCT/GB2003/003503
- (22) International Filing Date: 12 August 2003 (12.08.2003)
- (25) Filing Language: English
- (26) Publication Language: English
- (30) Priority Data:  
0219739.0 23 August 2002 (23.08.2002) GB  
0310076.5 1 May 2003 (01.05.2003) GB
- (71) Applicant (for all designated States except US): THE BOC GROUP PLC [GB/GB]; Chertsey Road, Windlesham, Surrey GU20 6HJ (GB).
- (72) Inventors; and  
(75) Inventors/Applicants (for US only): BROOKER, Brian, Edward [GB/GB]; "Masters", Thames Street, Sonning, Reading, Berkshire RG4 6UR (GB). TOMLINS, Richard, Ivor [GB/GB]; The Old Police House, Bix, Henley-on-Thames, Oxon RG9 6DE (GB).
- (74) Agent: WICKHAM, Michael; The BOC Group plc, Chertsey Road, Windlesham, Surrey GU20 6HJ (GB).
- (81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW.
- (84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PT, RO, SE, SI, SK, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).
- Published:**  
— with international search report  
— before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments
- For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.



WO 2004/017747 A1

(54) Title: MANUFACTURE OF ICE CREAM

(57) Abstract: Particles of edible fat are precrystallised at edible fat and are blended with an aqueous ice cream precursor phase in the presence of at least one emulsifier to form a dispersion. The precrystallised particles of edible fat each contain a multiplicity of individual crystals. The resulting dispersion is gasified and frozen so as to form an ice cream. The precrystallisation is preferably performed cryogenically.

# MANUFACTURE OF ICE CREAM

This invention relates to the manufacture of ice cream.

5 Ice cream is made by incorporating air into an oil-in-water emulsion as it is slowly frozen. By the term "ice cream" as used herein is meant a partly frozen foam containing a disperse phase of fat droplets and a continuous aqueous phase of dissolved and colloidal solids, such that the latter typically, but not essentially or exclusively, includes sugars (usually sucrose and lactose),  
10 proteins (usually milk solids) and stabilisers. Although the fat of choice is traditionally butterfat or dairy cream, increasing volumes of ice cream are made using hydrogenated vegetable or tropical hard fats. The disperse phase contains a mixture of oil and solid fat (ideally at least 75% solid or crystalline fat at -5°C) as well as an emulsifier (usually a mixture of lipophilic  
15 mono- and di-glycerides), whilst the aqueous phase contains a number of ingredients, of which milk protein, sugar and a stabiliser are the most important components in common use. Alternatively, a monoglyceride emulsifier may be dissolved in the aqueous phase with the other water-soluble ingredients.

20

In general, the manufacture of ice cream involves the following basic steps. After blending the selected liquid and dry ingredients using a high-speed mixer at 50-60°C, the coarse ice cream mix or emulsion is pasteurised, homogenised and cooled to about 4°C, by which time the droplets of oil/fat  
25 are stabilised by an interfacial layer of adsorbed milk proteins. Homogenisation of the emulsion involves pumping at high pressure through an homogenisation valve or orifice in order to produce a particle size reduction of the emulsion droplets. The cooled emulsion is held in "conditioning" or "ageing" tanks at about 5°C for some hours prior to freezing  
30 in order to allow a) the interfacial milk protein to be displaced from the surface of the fat globules by the more surface active emulsifier and b) the solid fat, especially the higher melting point fat, to crystallise. This "ageing" stage is

vitally important because it is only when the emulsifier has displaced the interfacial protein and itself comes to occupy the oil-water interface of each fat droplet and fat crystallisation has occurred, that it is possible for fat globules to stabilise air bubbles by attaching to their surface during air incorporation, and to thus form a stable, high volume foam for freezing. The time required for this "ageing" stage of production is long (and therefore expensive) and inevitably increases the cost of the final product, but it also dictates that the manufacture of ice cream is intrinsically a discontinuous process. This step is followed by air incorporation and dynamic freezing, both steps usually being performed in a scraped surface heat exchanger and, finally, hardening by blast freezing at  $-30^{\circ}$  to  $-40^{\circ}\text{C}$ .

According to the invention there is provided a method of making ice cream, including the steps of blending in the presence of at least one emulsifier an aqueous ice cream precursor phase with precrystallised particles of edible fat which each contain a multiplicity of individual crystals so as to form a dispersion, and gasifying and freezing the dispersion so as to form an ice cream.

By employing precrystallised particles of edible fat, the conventional homogenisation and ageing steps do not need to be performed. Furthermore, an emulsifier may be introduced into the edible fat before precrystallisation thereby enabling effective blending of the fat with the aqueous phase to be performed rapidly.

25

The particles of edible fat are preferably precrystallised cryogenically. The cryogenic precrystallisation may be performed by forming the edible fat into fine particles in molten state and contacting the fine particles with a cryogen. Typically, a spray of liquid cryogen is directed at the fine particles of edible fat in molten state. The liquid cryogen is conveniently liquid nitrogen, although liquid air or liquid argon may, for example be used instead.

30

The precrystallised particles of edible fat typically each take the form of a globule containing a mass of crystals (of very fine size) with entrapped pockets of oil.

- 5 Preferably, an emulsifier is introduced into the edible fat before it is precrystallised. The emulsifier is typically a lipophilic emulsifier, for example, a saturated monoglyceride such as glycerol monostearate.

- 10 Suitable classes of edible fat for use in the method according to the present invention include milk fat, anhydrous milk fat, at least one milk fat fraction, a hydrogenated vegetable oil, a hard tropical fat, or a hydrogenated tropical fat. Examples of non-dairy fats include fats formed by the hydrogenation of oils such as palm oil and sunflower oil.

- 15 The terms "cryogenic" and "cryogenically" as used herein mean that the temperature of the medium used to effect precrystallisation of the fat is less than minus 50°C, preferably less than minus 100°C.

- 20 One example of the method according to the invention involves cryogenically freezing a fine aerosol or spray or mist of a suitable fat (that can for example be milk fat, anhydrous milk fat, one or more milk fat fractions, hydrogenated vegetable oil, a tropical hard fat or any other edible oil containing a suitable level of solid fat) that may contain an appropriate added emulsifier, to produce a fine fat powder, although other cooling processes that achieve these
- 25 conditions may also be used for the same purpose. When the fat powder is returned to ambient temperature, or the temperature of the ice cream making process, it consists of droplets of fat, or fat globules, in which crystallisation of the fat is complete and the emulsifier is associated with their surface. Addition of this powder to the pasteurised aqueous phase of the ice cream
- 30 formulation, followed by high speed mixing, produces a mixture that is immediately ready for air (or other gas) incorporation and freezing in a

conventional manner and therefore the homogenisation and "conditioning" or "ageing" stages are no longer necessary.

In a further example a lipophilic emulsifier such as a saturated monoglyceride, for example, glycerol monostearate is dispersed in a molten, commercial ice cream fat and the fat then sprayed as a fine aerosol or as a mist from a spinning disc or some other similar device that produces small droplets, to impinge onto a liquid cryogen, or co-sprayed with the cryogen, or in some other way brought into close contact with a cryogen to obtain the highest possible rate of cooling. The resulting fine powder consists of numerous spherical fat particles, each of which consists of a mass of very fine fat crystals with entrapped pockets of oil. These particles can be added directly to the aqueous phase of ice cream mixes to produce a fine fat dispersion of what is, in effect, an emulsion. The fat powder can be wetted by the aqueous phase because of the presence of emulsifier on the surface of each powder particle. Individual fat particles act as globules of fat that are able to stabilise air bubbles by attaching to their surface when, as part of the normal ice cream making process, air is incorporated into the aqueous phase during freezing. Emulsions formed in this way freeze to form ice creams with similar microscopic structure, physical properties and mouth-feel as ice creams prepared with the same materials using conventional methods.

In a yet further example of this invention, a lipophilic emulsifier is dissolved in the fat phase, as described above, but the aqueous phase to which the fat is added also contains a highly surface active, water soluble emulsifier, such as a Tween or a Span (for example, Polysorbate 60 or Span 60. Whereas the lipophilic emulsifier water facilitates wetting of the fat particles' surface by the aqueous phase during blending and ensures that emulsifier is already present at the surface of the fat particles when the aqueous phase is added, the water soluble emulsifier lowers the surface tension of the fat particles to very low levels and promotes the separation (and wetting) of fat crystals from fat particles; in this way, large numbers of emulsifier-coated fat crystals become

dispersed in the aqueous phase. Thus, both fat crystals and fat particles are available to attach to the surface of bubbles and to take part in air (or other gas) stabilisation when the aeration process begins, to produce a stable foam with high overrun, that is the increase in volume produced by air incorporation, and a distinctive mouth-feel in the frozen ice cream.

The method of the invention has the advantage that it provides a number of ways of preparing and using the fat phase in an ice cream mix. Such is the flexibility of the method of the invention that it has also been found possible to crystallise the fat phase without an added emulsifier and instead to associate an emulsifier with the surface of the fat droplets by including the surfactant solely in the aqueous phase of the ice cream formulation. In this case, the emulsifier becomes associated with the particles' surface at the beginning of the ice cream making process during the high speed blending of the ingredients.

A positive advantage of the method according to the invention is that the disperse fat phase (i.e. the precrystallised particles of edible fat) can be prepared and conveniently stored as a stable powder until it is needed and then added either a) to the prepared aqueous phase or b) to other, dry ingredients of the ice cream mix followed by the addition of water and high speed mixing. This instant formation of an oil-in-water ice cream mix ready for immediate aeration and freezing provides a simple, rapid and continuous process for ice cream production that avoids many of the problems, delays and the inherent costs encountered in conventional, batch processes.

The method of the invention can be effectively placed upstream of the freezing stage of a continuous ice cream making plant so that holding tanks or "re-crystallisers" in which the emulsion would normally be aged, are no longer necessary in the process. This is possible using the method of the invention because within each droplet, fat crystallisation is rapid and complete and can be used at once in ice cream making or stored for extended periods without

adverse changes in appearance or functionality. The stability of fat powders produced by the method of the invention can be explained by the fact that all of the fat dissolved in the oil phase at the beginning of the process is forced to crystallise during cryogenic treatment. This is not true of other methods already in use for the production of fat powders in which there is only partial crystallisation of the fat by the end of the process and a tempering or storage stage is required to allow the process to reach equilibrium – often with the release of sufficient heat to cause partial melting and subsequent lumping or caking. Such powders are intrinsically unstable when exposed to temperature fluctuations because some solid fat is still present in solution in the oil phase and this allows tempering effects to change the properties of the fat crystals.

In view of the stability of the cryogenically precrystallised edible fat, it is possible to package it by itself or with one or more other solid constituents of an ice cream mix for use in the home preparation of ice cream. Instructions on the use of the contents of the package to make ice cream can be provided as part of a kit with the package.

Moreover, an additional benefit of the method of the invention is that the exit temperature of the frozen fat powder can be adjusted within certain limits (depending on the cryogen used) so that after it has been mixed with the aqueous phase, the complete emulsion can enter the freezing stage in a pre-cooled state, i.e. at a sub-ambient temperature (e.g. in the range of 10 to 15°C). This increases the efficiency of operation of ice cream making by utilising some of the energy used in the production of the cryogenic gas.

Two further advantages of the method according to the invention are that in cryogenically precrystallised particles of fat powders a) the fat crystal size is very much smaller than in fat powders prepared by spray drying or spray chilling and b) the particles have an elevated solid fat content compared with the same fat prepared by conventional methods. Whereas the small crystal size can be produced in all edible fats that are precrystallised cryogenically,

the increase in solid fat content is especially notable in cases where the fat to be cryogenically treated contains or consists of a hydrogenated fat or a fraction or blend containing triglycerides with a significant percentage of saturated fatty acids, such as hydrogenated fats and tropical hard fats. The increase in solid fat content of fat powders produced by the method of the invention increases their functionality when used in a wide range of applications, from baking to ice cream making. This cannot be achieved using existing, conventional methods of fat powder manufacture.

- 10 In the method of the invention it has been found that the size of particles in the fat powder is influenced by a number of factors that are already known, including typically a) the design, specification and operating conditions of the aerosol nozzle or of the rotary atomiser or of the device used to produce the fine dispersion of fat to be frozen by the cryogen and b) the viscosity of the
- 15 molten fat at the atomiser feed temperature, whilst edible fat particle size in the aqueous phase of the ice cream mix prior to freezing is influenced by a) the magnitude of the shear applied in dispersing the fat powder in the aqueous phase of the ice cream mix and b) the solid fat content of the fat powder at the temperature of high energy mixing with the aqueous phase, in
- 20 accordance with established principles of emulsion behaviour. In general, a high speed mixer is employed to perform the blending step in the method according to the invention.

- It has been found advantageous to freeze an ice cream mix (i.e. the dispersion) in which all the dispersed fat particles in the dispersion have a
- 25 size or diameter of less than  $30\mu\text{m}$ , preferably less than  $10\mu\text{m}$  and optimally no more than  $5\mu\text{m}$  in diameter, since particles larger than  $25\text{-}30\mu\text{m}$  in diameter are known to produce poor mouth-feel, low overrun and loss of quality.



The edible fat to be used in the method according to the invention is, preferably, one with proven use in ice cream making and whose solid fat content at the temperature of storage is high enough to maintain its state as a non-compactable, free-flowing powder. This has practical advantages in the automated delivery of ingredients to unit operations, such as mixing. It has been found by experiment that powders made from high melting point butter fractions, a number of hard tropical fats, hydrogenated tropical fats, as well as a number of hydrogenated vegetable oils satisfy these conditions at storage temperatures below 10°C but that butter, anhydrous butter and lower melting point butter fractions must ideally be stored near to or below 0°C (depending on their triglyceride composition) to remain free flowing.

The microbiological quality of the final fat powder can be controlled by pasteurising the molten fat before cryogenic re-crystallisation. This can be done by simply adjusting the temperature of the molten fat in the holding tank.

The method according to the invention is further illustrated by the following examples.

#### EXAMPLE 1

An ice cream mix was prepared by combining a powdered precrystallised non-dairy fat, produced according to the method of the invention, with a pre-prepared aqueous phase to give a final fat content of 9%.

The fat powder was prepared by melting a commercial ice cream fat that was composed of hardened palm kernel oil containing 0.5% of a purified monoglyceride derived from sunflower oil, followed by cryogenic spray crystallisation using liquid nitrogen. The mean particle size was 70µm. The solid fat content of the starting material was determined using pulsed NMR over a narrow range of temperatures (-5°C, 91.1%; 0°C, 88.6%; +5°C, 83.5%).

The aqueous phase, which had the composition given in Table 1 below, was prepared using a low speed mixer.

**Table 1. Composition of the aqueous phase of an ice cream mix.**  
**(% by weight final ice cream mix)**

5	Non-fat dry milk solids	11.5%
	Sugar	14.0%
	Hydrocolloid stabiliser	0.3%
10	Water	65.2%

The edible fat powder and the aqueous phase were thoroughly mixed at a temperature within the range 10°C-15°C using a high speed blender and the resulting dispersion or ice cream mix was aerated and frozen in a Votator (Trade Mark) continuous freezer. The emergent ice cream was hardened by passage through a tunnel freezer and the resulting material stored at -20°C for 2 days before evaluation. If desired, the emergent ice cream can be packaged before having hardened.

20 This ice cream was compared with a control ice cream prepared from the same materials but produced by a conventional process in which the fat phase was dispersed by homogenisation and the resulting ice cream mix "aged" for several hours before freezing as described above.

25 Comparison was made of the two ice creams by measurement of their overrun and yield values, the latter being a measure of the hardness of the ice cream. The former was measured using standard methods and the latter by means of an Instron penetrometer fitted with a refrigerated sample holder and a 40° cone probe. Organoleptic observations were also noted to try to detect  
30 any obvious defects in ice cream structure.

These comparisons, set out in Table 2 below, showed that the properties of the two ice creams were similar.

**Table 2. Comparison of the overrun, yield value and organoleptic properties of two ice creams.**

Method of ice cream Preparation	Overrun % (SD=standard deviation)	Average Yield Value (n= 5) dynes/cm <sup>2</sup> x10 <sup>4</sup>	Organoleptic observations
Combining cryogenically processed fat powder and aqueous phase	53 SD 3.9	5.5	Smooth
Control – conventional homogenisation + “ageing”	60 SD 3.4	5.2	Very smooth

## EXAMPLE 2

A dispersion of an ice cream mix was prepared by combining a cryogenically precrystallised anhydrous milk fat (AMF) powder, with a pre-prepared aqueous phase to give a final fat content of 12.5%.

- 15 The AMF powder was prepared by melting a commercial AMF and adding a commercial monoglyceride (glycerol monostearate) to a level of 0.3%, followed by cryogenic spray crystallisation using liquid nitrogen. The solid fat content of the AMF (before addition of the monoglyceride) was determined using pulsed NMR over a narrow range of temperatures (-5°C, 70.7%; 0°C,
- 20 67.5%; +5°C, 60.0%). The aqueous phase, which had the composition given in Table 3 below, was prepared using a low speed mixer.

**Table 3. Composition of the aqueous phase used to make a dairy ice cream.**

**(% by weight final ice cream mix)**

5	Non-fat dry milk solids	13.0%
	Sugar	12.0%
	Commercial hydrocolloid stabiliser	0.2%
	Water	62.3%

10 The AMF powder and the aqueous phase were blended in a high speed mixer and the resulting ice cream mix was aerated and frozen in a Votator continuous freezer. The emergent dairy ice cream was hardened in a tunnel freezer and the resulting material stored at  $-20^{\circ}\text{C}$  for 2 days before evaluation.

15 This ice cream was compared with a control ice cream prepared from the same materials but produced by a conventional process in which the AMF was dispersed by homogenisation and the resulting ice cream mix "aged" for several hours before freezing as described above.

20 Comparison was made of the two ice creams by measurement of their overrun and yield values, as in Example 1. Organoleptic observations were also noted to try to detect any obvious defects in ice cream structure. The results of the comparison are set out in Table 4 below.

25 The results showed that there was little difference in the physical properties of the two ice creams but that the mouthfeel of the product made conventionally was probably smoother than would be acceptable commercially.

30

**Table 4. Comparison of the overrun, yield value and organoleptic properties of two dairy ice creams.**

<b>Method of ice cream Preparation</b>	<b>Overrun % (SD=standard deviation)</b>	<b>Average Yield Value (n= 5) Dynes/cm<sup>2</sup>x10<sup>4</sup></b>	<b>Organoleptic observations</b>
Combining cryogenically processed AMF powder and aqueous phase	58 SD 4.8	5.1	Very smooth
Control – conventional homogenisation + “ageing”	67 SD 4.9	4.8	Very smooth to too smooth (“slimy”)

### 5 EXAMPLE 3

A dispersion of an ice cream mix was prepared by combining a powdered non-dairy fat, produced by cryogenic precrystallisation with a pre-prepared aqueous phase to give a final fat content of 10%.

10

The fat powder was prepared by melting a palm kernel oil containing 0.3% of a purified monoglyceride derived from sunflower oil, followed by cryogenic spray crystallisation using liquid nitrogen. The cryogenically precrystallised fat powder has a mean particle size of 70µm. The solid fat content of the starting material was determined using pulsed NMR over a narrow range of temperatures (-5°C, 86.2%; 0°C, 84.3%; +5°C, 78.9%). The aqueous phase, which had the composition given in Table 5 below, was prepared using a low speed mixer.

15

**Table 5. Composition of the aqueous phase of an ice cream mix containing polysorbate 60 (% by weight final ice cream mix).**

5	Non-fat dry milk solids	11.5%
	Sugar	14.0%
	Hydrocolloid stabiliser	0.3%
	Polysorbate 60	0.1%
10	Water	64.1%

The fat powder and aqueous phase were thoroughly mixed using a high speed blender and the resulting ice cream mix was aerated and frozen in a Votator continuous freezer. The semi-solid ice cream that emerged was  
 15 hardened by passage through a tunnel freezer and the resulting material stored at  $-20^{\circ}\text{C}$  for 2 days before evaluation.

This ice cream was compared with a control ice cream prepared from the same materials but produced by a conventional process in which the fat  
 20 phase was dispersed by homogenisation and the resulting ice cream mix "aged" for several hours before freezing as described above.

Comparison was made of the two ice creams using measurements as described in examples 1 and 2 and organoleptic observations were also noted  
 25 to detect any obvious defects in ice cream structure. The results of the comparison are set out in Table 6 below. These showed that the two ice creams were similar in their properties.

**Table 6. Comparison of the overrun, yield value and organoleptic properties of two ice creams prepared with two added emulsifiers.**

30

Method of ice cream Preparation	Overrun % (SD=standard deviation)	Average Yield Value (n= 5) dynes/cm <sup>2</sup> x10 <sup>4</sup>	Organoleptic observations
Combining cryogenically processed fat powder and aqueous phase	70 SD 3.5	4.6	Very smooth
Control – conventional homogenisation + “ageing”	75 SD 3.0	4.2	Very smooth

## CLAIMS

1. A method of making ice cream, including the steps of blending in the presence of at least one emulsifier an aqueous ice cream precursor  
5 phase with precrystallised particles of edible fat which each contain a multiplicity of individual crystals so as to form a dispersion, and gasifying and freezing the dispersion so as to form an ice cream.
2. A method according to claim 1, in which the particles of edible fat are  
10 precrystallised cryogenically.
3. A method according to claim 2, in which the cryogenic precrystallisation is performed by forming the edible fat into fine particles in molten state and contacting the fine particles with a cryogen.  
15
4. A method according to claim 3, in which a spray of liquid cryogen is directed at the fine particles of edible fat in molten state.
5. A method according to claim 3 or claim 4, in which the liquid cryogen is  
20 liquid nitrogen.
6. A method according to any one of the preceding claims, in which the precrystallised particles of edible fat take the form of a globule containing a mass of crystals of fat with entrapped pockets of oil.  
25
7. A method according to any one of the preceding claims, in which all the dispersed fat particles in the dispersion have a size less than 30µm.
8. A method according to claim 7, in which most or all the precrystallised  
30 particles have a size less than 10µm.



9. A method according to claim 7 or claim 8, in which most or all of the precrystallised particles have a size of 5µm or less.
- 5 10. A method according to any one of the preceding claims, in which the edible fat is pasteurised before being precrystallised.
11. A method according to any one of the preceding claims, in which the aqueous phase is pasteurised before being blended with the precrystallised edible fat particles.
- 10 12. A method according to any one of the preceding claims, in which an emulsifier is introduced into the edible fat before it is precrystallised.
13. A method according to claim 12, in which the emulsifier is a lipophilic emulsifier.
- 15 14. A method according to claim 13, in which the lipophilic emulsifier is a saturated monoglyceride.
- 20 15. A method according to claim 14, in which the saturated monoglyceride is a glycerol monostearate.
16. A method according to any one of the preceding claims, in which the edible fat is milk fat, anhydrous milk fat, at least one milk fat fraction, a hydrogenated vegetable oil, a hard tropical fat, or a hydrogenated tropical fat.
- 25 17. A method according to any one of the preceding claims, in which the aqueous phase contains a highly surface active, water soluble emulsifier.
- 30

18. A method according to any one of the preceding claims, in which the aqueous phase contains non-fat dry milk solids and sugar.
- 5 19. A method according to any one of the preceding claims, in which the said dispersion is gasified and frozen without being subjected to homogenisation or ageing.
20. A method according to claim 19, in which the dispersion is presented at below ambient temperature for freezing.
- 10 21. A package comprising cryogenically precrystallised particles of edible fat.
- 15 22. A kit for making ice cream in the home comprising a package according to claim 21 and instructions for the use of the contents of the package in the preparation of ice cream.

## **ABSTRACT**

### **MANUFACTURE OF ICE CREAM**

5

Particles of edible fat are precrystallised at edible fat and are blended with an aqueous ice cream precursor phase in the presence of at least one emulsifier to form a dispersion. The precrystallised particles of edible fat each contain a multiplicity of individual crystals. The resulting dispersion is gasified and

10

frozen so as to form an ice cream. The precrystallisation is preferably performed cryogenically.

# INTERNATIONAL SEARCH REPORT

International Application No.  
PCT/GB 03/03503

## A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 A23G9/02 A23G3/00 A23G9/04 A23G9/20

According to International Patent Classification (IPC) or to both national classification and IPC

## B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 A23G

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the International search (name of data base and, where practical, search terms used)

EPO-Internal, WPI Data, PAJ

## C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO 02 49445 A (BOURKE NEIL JOSEPH ;QUEST INTERNAT B V (NL)) 27 June 2002 (2002-06-27) page 5, line 1 - line 30; claims; examples 9,2,3,4 page 6, line 7 - line 17 ---	1,10-21
X	US 6 349 549 B1 (ANGUS NICHOLAS W ET AL) 26 February 2002 (2002-02-26) column 2, line 64 -column 3, line 20; figures 5,7 column 1, line 6 - line 8 column 3, line 54 -column 4, line 4 ---	1,2,21
Y	EP 0 147 483 A (PILLSBURY CO) 10 July 1985 (1985-07-10) claims; examples ---	1-5,21
	-/--	

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

### \* Special categories of cited documents:

- \*A\* document defining the general state of the art which is not considered to be of particular relevance
- \*E\* earlier document but published on or after the International filing date
- \*L\* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)
- \*O\* document referring to an oral disclosure, use, exhibition or other means
- \*P\* document published prior to the International filing date but later than the priority date claimed

- \*T\* later document published after the International filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention
- \*X\* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone
- \*Y\* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.
- \*G\* document member of the same patent family

Date of the actual completion of the International search

14 November 2003

Date of mailing of the International search report

28/01/2004

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2  
NL - 2280 HV Rijswijk  
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,  
Fax: (+31-70) 340-3016

Authorized officer

Guyon, R

# INTERNATIONAL SEARCH REPORT

Intern. Application No.

PCT/GB 03/03503

## C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
Y	WO 95 20324 A (GRATED CHEESE COMPANY LIMITED ; SMITH MALYON MCCREA (NZ)) 3 August 1995 (1995-08-03) figures ---	1-5, 21
Y	US 4 434 186 A (BROUWER PETER ET AL) 28 February 1984 (1984-02-28) claims 1, 6, 13 ---	1
Y	EP 0 289 069 A (ASAHI CHEMICAL IND) 2 November 1988 (1988-11-02) column 1, line 37 - line 40 column 2, line 38 - line 49 ---	1
A	EP 0 727 148 A (UVIGAL SPA) 21 August 1996 (1996-08-21) the whole document ---	1, 11, 22
A	US 4 219 581 A (DEA IAIN C M ET AL) 26 August 1980 (1980-08-26) ---	
A	EP 1 052 284 A (TAIYO KAGAKU KK) 15 November 2000 (2000-11-15) -----	

# INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/GB 03/03503

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
WO 0249445	A	27-06-2002	AU 2552702 A BR 0116454 A DK 200300922 A EP 1351579 A1 WO 0249445 A1	01-07-2002 07-10-2003 18-07-2003 15-10-2003 27-06-2002
US 6349549	B1	26-02-2002	US 6574969 B1 WO 02070969 A1	10-06-2003 12-09-2002
EP 0147483	A	10-07-1985	EP 0147483 A1	10-07-1985
WO 9520324	A	03-08-1995	AU 685906 B2 AU 1547295 A BR 9506584 A GB 2300105 A ,B HK 1008414 A1 JP 3218042 B2 JP 9508278 T WO 9520324 A1 NZ 278979 A	29-01-1998 15-08-1995 30-09-1997 30-10-1996 07-05-1999 15-10-2001 26-08-1997 03-08-1995 24-10-1997
US 4434186	A	28-02-1984	NONE	
EP 0289069	A	02-11-1988	JP 1027430 A JP 2601300 B2 AU 626790 B2 AU 1431588 A DE 3879748 D1 DE 3879748 T2 EP 0289069 A2 US 5127953 A US 5043018 A	30-01-1989 16-04-1997 13-08-1992 06-10-1988 06-05-1993 28-10-1993 02-11-1988 07-07-1992 27-08-1991
EP 0727148	A	21-08-1996	IT MI950283 A1 EP 0727148 A2	16-08-1996 21-08-1996
US 4219581	A	26-08-1980	AT 375817 B AT 306879 A AU 532829 B2 AU 4640979 A BE 875826 A1 CA 1117355 A1 CH 652567 A5 DE 2916395 A1 DK 166879 A ES 479892 A1 FR 2423986 A1 GB 2019187 A ,B IE 48114 B1 IT 1118967 B LU 81186 A1 NL 7903236 A PT 69543 A SE 7903559 A ZA 7901935 A	10-09-1984 15-02-1984 13-10-1983 01-11-1979 24-10-1979 02-02-1982 29-11-1985 31-10-1979 25-10-1979 16-08-1980 23-11-1979 31-10-1979 03-10-1984 03-03-1986 07-11-1979 26-10-1979 01-05-1979 25-10-1979 26-11-1980
EP 1052284	A	15-11-2000	JP 2000119687 A JP 2000116333 A	25-04-2000 25-04-2000

# INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/GB 03/03503

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 1052284	A	JP 2000116349 A	25-04-2000
		JP 2000116330 A	25-04-2000
		JP 2000116322 A	25-04-2000
		JP 2000116323 A	25-04-2000
		JP 2000116325 A	25-04-2000
		JP 2000119688 A	25-04-2000
		JP 2000116357 A	25-04-2000
		AU 8463298 A	23-08-1999
		CA 2319425 A1	12-08-1999
		EP 1052284 A1	15-11-2000
		US 6346289 B1	12-02-2002
		WO 9940167 A1	12-08-1999